

# Voyage in the Electronic Aspects of Phosphorus Chemistry

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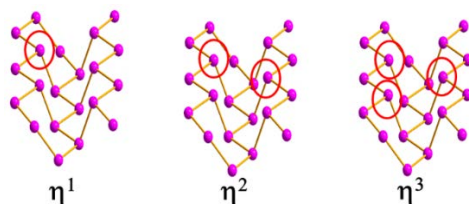
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Phosphorus shares with carbon the possibility of various allotropes with different stability and structure. Unlike white phosphorus, P<sub>4</sub>, formed by discrete tetrahedral molecules, is highly reactive, the black allotrope (bP) one shows a layered extended structure and exhibits the most thermodynamic stability. Some years ago, the exfoliation the bP provides the mono-layer 2D material phosphorene (P<sub>n</sub>),<sup>[1]</sup> which, due to its fast reactivity with oxygen, has to be protected or functionalized. Herein, we present a voyage in the structural/electronic underpinnings, which rules the reactivity of the phosphorus allotropes with small molecules, such as di-iodine, and transition metal fragments. In particular, we started with the P<sub>4</sub> for which a detailed analysis of the reaction between P<sub>4</sub> and I<sub>2</sub> highlighted multi-step concerted process with of 3 + 3 bond breakings/makings in each case.<sup>[2]</sup> On the gathered information on the simplest P<sub>4</sub>, we explored the plausibility of some reaction pathway of P<sub>n</sub> through solid state DFT calculations. The high density of the facial P atoms, with outpointing but not fully independent lone pairs, offers potential P<sub>n</sub> reactivity with mono, bi- and three-functional acidic units, see Figure 1. In particular, we examined the reliability of the adducts between P<sub>n</sub> and BH<sub>3</sub>, I<sub>2</sub> or the ClAu(I) fragment.<sup>[3]</sup> Other unsaturated Transition Metal fragments of the L<sub>2</sub>M and L<sub>3</sub>M type were chosen on the basis of the *isolobal analogy* concept<sup>[4]</sup> for combining neighbor P<sub>n</sub> atoms with a single metal that carries multiple vacant lobes.



**Fig.1.** Different potential coordination of acids or metal fragments on the P<sub>n</sub> surface.

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